

**Department of Pure and Applied Chemistry** 

University of Kota, Kota

# **M.Sc. III Semester Chemistry**

# Lecture Plans for Academic Session 2018-19

# **Common Papers for All Specializations**

## Paper-3.1: CHEM-631: Chromatography

S.	Unit wise Broad Topics of the Syllabus	Tentative
No.		Lecture(s) Allotted
	I: General Introduction of Separation:	
1.	Nature of separation process,	1
2.	classification of separation methods	
	Chromatography:	
3.	General introduction, principles and types of chromatography,	2-3
4.	physical sate of mobile phase,	4
5.	mechanism of separation and techniques involved	5
	Paper Chromatography:	
6.	Principle, types, choice of paper and solvent,	6-8
7.	location of spot, development, visualization, measurement of R <sub>f</sub> values,	9-11
8.	applications	12
	Supercritical Fluid Chromatography (SFC):	
9.	Principle, instrumentation, qualitative and quantitative analysis	13-15
Unit	II: Thin Layer Chromatography (TLC):	
10.	Principle, advantage over paper chromatography,	16
11.	types,	17
12.	preparation of thin layer,	18
13.	choice of sorbent and solvent,	19
14.	development, detection	20
15.	applications	21
	High Performance Thin Layer Chromatography (HPTLC):	
16.	Principle, advantage over TLC,	22
17.	instrumentation,	23-25
18.	choice of sorbent and solvent,	26-27
19.	development, detection	28-29
20.	applications	30
Unit	III: Column Chromatography:	
21.	Principle, resolution, stationery phase,	31
22.	column efficiency,	32-33
23.	factors influencing column efficiency,	34-35
24.	experimental set up and applications;	36
25.	principle and application of flash chromatography	37
	Gas Chromatography (GC):	
26.	Principle, instrumentation,	38
27.	column efficiency,	39
28.	solid supports, liquid phase,	40
29.	column temperature,	41

30.	detectors,	42-43
31.	chromatographic identification,	44
32.	multi-dimensional GC, fast GC,	
33.	applications	45
Unit	-IV: High Performance Liquid Chromatography (HPLC):	
34.	Principle, instrumentation,	46
35.	identification of peaks,	47
36.	effect of temperature and packing material,	
37.	types of HPLC: partition, adsorption, ion-exchange, size-exclusion or gel;	48-49
38.	derivatization in HPLC: post and pre-columns,	50
39.	applications	51
	Ion-Exchange or Ion Chromatography (IC):	
40.	Principle, types, regeneration,	52
41.	ion-exchange resins and their capacity,	53-55
42.	retention, selectivity,	56
43.	factors affecting separation,	57
44.	bonded phase chromatography (BPC),	58
45.	high performance ion chromatography (HPIC),	59
46.	applications	60
Unit	-V: Electrophoresis:	
47.	Theory and classification,	61
48.	factors affecting mobility,	62
49.	macromolecular size and charge interactions with supporting electrolyte, pH and concentration discontinuities, factors affecting,	63-64
50.	electrophoresis phenomena: electrolysis, electro-osmosis, temperature and supporting media;	65-66
51.	instrumentation, methodology,	67
52.	preparation of gel-staining and de-staining,	68
53.	preparative zone electrophoresis,	69
54.	continuous electrophoresis,	
55.	applications	70
	Capillary Electrophoresis (CE):	
56.	Principle, theory,	71
57.	instrumentation,	72
58.	sample preparation and applications,	73
59.	capillary electro-chromatography	74
60.	miscellar electro-kinetic capillary chromatography	75

- Chromatography: Basic Principles, Sample Preparations and Related Methods by Elsa Lundanes, Leon Reubsaet, Tyge Greibrokk, John Wiley and Sons
- Introduction to Modern Liquid Chromatography by Lloyd R. Snyder, Joseph J. Kirkland and John W. Dolan, Wiley
- Practical HPLC Method Development by Lloyd R. Snyder, Wiley-Interscience
- Principles & Practices of Chromatography by R. P. W. Scott, Library for Science
- Fundamentals of Analytical Chemistry, VIII Edn., D. A. Skoog, D. M. West, F.J. Holler and S.R. Crouch, Thomson Brooks/Cole Publishers.
- Principles of Instrumental Analysis by D.A. Skoog, F.J. Holler and T.A. Nieman, 5th Edition, Harcourt Brace & Company, Florida.
- Instrumental Methods of Chemical Analysis, B. K. Sharma, Goel Publishing House, Meerut.
- Instrumental Methods of Chemical Analysis, Chatwal and Anand, Himalaya Publishing House, Meerut.
- Basic Gas Chromatography 2nd Edition by Harold M. McNair, James M. Miller, John Wiley and Sons.

- Comprehensive two dimensional gas chromatography, Volume 55 (Comprehensive Analytical Chemistry) by Lourdes Ramos, Elsevier
- Forensic Applications of Gas Chromatography 1st Edition by Michelle Groves Carlin, John Richard Dean, Taylor & Francis
- Analytical Gas Chromatography 2nd Edition by Phillip Stremple, Elsevier
- Electrophoresis by Duncan J. Shaw. Academic Press
- Gel Electrophoresis-Advanced Techniques Edited by Sameh Magdeldin. InTech.
- Capillary Electrophoresis Guidebook: Principles, Operation, and Applications by Kevin D. Altria. Springer Science & Business Media.

Unit-I: Ultraviolet-Visible (UV-VIS) Spectroscopy :1.Electromagnetic radiation and spectroscopy,2.principles of absorption spectroscopy,3.nature of electronic excitations,4.chromophores, auxochromes,5.origin of UV bands,6.absorption and intensity shifts,7.types of absorption bands,8.factors affecting the position of UV bands,9.calculation of $\lambda_{max}$ of simple organic compounds,10.visible spectra,11.qualitative and quantitative applicationsInfrared (IR) Spectroscopy:12.IR regions,13.molecular vibrations,14.force constant and bond strengths,15.calculation of vibrational frequencies,	
1.Electromagnetic radiation and spectroscopy,2.principles of absorption spectroscopy,3.nature of electronic excitations,4.chromophores, auxochromes,5.origin of UV bands,6.absorption and intensity shifts,7.types of absorption bands,8.factors affecting the position of UV bands,9.calculation of $\lambda_{max}$ of simple organic compounds,10.visible spectra,11.qualitative and quantitative applications12.IR regions,13.molecular vibrations,14.force constant and bond strengths,	2
2.principles of absorption spectroscopy,3.nature of electronic excitations,4.chromophores, auxochromes,5.origin of UV bands,6.absorption and intensity shifts,7.types of absorption bands,8.factors affecting the position of UV bands,9.calculation of $\lambda_{max}$ of simple organic compounds,10.visible spectra,11.qualitative and quantitative applications <b>Infrared (IR) Spectroscopy:</b> 12.IR regions,13.molecular vibrations,14.force constant and bond strengths,	3
3.       nature of electronic excitations,         4.       chromophores, auxochromes,         5.       origin of UV bands,         6.       absorption and intensity shifts,         7.       types of absorption bands,         8.       factors affecting the position of UV bands,         9.       calculation of $\lambda_{max}$ of simple organic compounds,         10.       visible spectra,         11.       qualitative and quantitative applications         Infrared (IR) Spectroscopy:         12.       IR regions,         13.       molecular vibrations,         14.       force constant and bond strengths,	3
5.       origin of UV bands,         6.       absorption and intensity shifts,         7.       types of absorption bands,         8.       factors affecting the position of UV bands,         9.       calculation of $\lambda_{max}$ of simple organic compounds,         10.       visible spectra,         11.       qualitative and quantitative applications         Infrared (IR) Spectroscopy:         12.       IR regions,         13.       molecular vibrations,         14.       force constant and bond strengths,	
5.       origin of UV bands,         6.       absorption and intensity shifts,         7.       types of absorption bands,         8.       factors affecting the position of UV bands,         9.       calculation of $\lambda_{max}$ of simple organic compounds,         10.       visible spectra,         11.       qualitative and quantitative applications         Infrared (IR) Spectroscopy:         12.       IR regions,         13.       molecular vibrations,         14.       force constant and bond strengths,	
6.absorption and intensity shifts,7.types of absorption bands,8.factors affecting the position of UV bands,9.calculation of $\lambda_{max}$ of simple organic compounds,10.visible spectra,11.qualitative and quantitative applicationsInfrared (IR) Spectroscopy:12.IR regions,13.molecular vibrations,14.force constant and bond strengths,	
7.       types of absorption bands,         8.       factors affecting the position of UV bands,         9.       calculation of $\lambda_{max}$ of simple organic compounds,         10.       visible spectra,         11.       qualitative and quantitative applications         Infrared (IR) Spectroscopy:         12.       IR regions,         13.       molecular vibrations,         14.       force constant and bond strengths,	4
8.       factors affecting the position of UV bands,         9.       calculation of $\lambda_{max}$ of simple organic compounds,         10.       visible spectra,         11.       qualitative and quantitative applications         Infrared (IR) Spectroscopy:         12.       IR regions,         13.       molecular vibrations,         14.       force constant and bond strengths,	4
9.       calculation of $\lambda_{max}$ of simple organic compounds,         10.       visible spectra,         11.       qualitative and quantitative applications         Infrared (IR) Spectroscopy:         12.       IR regions,         13.       molecular vibrations,         14.       force constant and bond strengths,	
<ul> <li>10. visible spectra,</li> <li>11. qualitative and quantitative applications</li> <li>Infrared (IR) Spectroscopy:</li> <li>12. IR regions,</li> <li>13. molecular vibrations,</li> <li>14. force constant and bond strengths,</li> </ul>	5-7
<ol> <li>qualitative and quantitative applications</li> <li>Infrared (IR) Spectroscopy:</li> <li>IR regions,</li> <li>molecular vibrations,</li> <li>force constant and bond strengths,</li> </ol>	8
Infrared (IR) Spectroscopy:         12.       IR regions,         13.       molecular vibrations,         14.       force constant and bond strengths,	
12.IR regions,13.molecular vibrations,14.force constant and bond strengths,	
<ol> <li>molecular vibrations,</li> <li>force constant and bond strengths,</li> </ol>	9
14. force constant and bond strengths,	
U ·	10
16. Fermi resonance, combination bands, overtones, hot bands,	11
17. factors affecting the band positions and intensities,	12
18. sample handling,	13
19. anharmonicity,	
20. group frequencies,	14
21. applications	15
Unit-II: Nuclear Magnetic Resonance (NMR) Spectroscopy:	
22. Nuclear angular momentum, nuclear spin,	16
23.     magnetization & nuclear precession,	10
24. free induction decay,	17
25. population densities of nuclear spin states,	
26. types of NMR spectrometers,	18
27. basic theory,	
28. equivalence & non-equivalence protons,	19
29. shielding and de-shielding of nuclei,	
30. chemical shift and its measurements,	20
31.     factors influencing chemical shift.	
32. spin-spin interactions: theory, types, factors influencing coupling constant "J".	21-22
<ul> <li>33. typical <sup>1</sup>H NMR absorption signals of various type of compounds.</li> </ul>	
34.     spin systems & classification of spectra	23

35.	splitting patterns of AX, ABX, AMX, ABC, A <sub>2</sub> B <sub>2</sub> , <i>etc.</i> spin systems	24-26
36.	simplification of spectra: shift reagents and spin decoupling;	27-20
30. 37.	proton exchange,	28
38.	nuclear overhauser effect,	20
<u>39.</u>	basic idea about NMR studies of nuclei other than proton <i>viz</i> . <sup>11</sup> B, <sup>15</sup> N, <sup>19</sup> F & <sup>31</sup> P	29
<del>1</del> 0.	applications of NMR spectroscopy	30
	III: Carbon-13 NMR Spectroscopy:	30
41.	Carbon-13 nucleus, operating frequency,	31
12.	carbon-13 include, operating nequency, carbon-13 chemical shifts and their calculation,	31
+2. 13.	spin-spin coupling: proton-coupled, proton-decoupled and off-resonance carbon-	33-34
+3.	13 spectra	55-54
14.	applications of <sup>13</sup> C NMR spectroscopy	35
17.	Electron Spin Resonance (ESR) Spectroscopy:	55
45.	Basic principles,	36
46.	zero field splitting and Kramer's degeneracy,	37
47.	factors affecting the 'g' value,	38
+7. 48.	hyperfine splitting,	39-40
49.	isotropic and anisotropic hyperfine coupling constants,	41
<del>19.</del> 50.	spin-orbit coupling,	41
50. 51.	significance of g-tensor,	43
51. 52.	spin Hamiltonian, spin densities and McConnell relationship,	44
52. 53.	measurement techniques and applications	45
	IV: Mass Spectrometry :	
54.	Basic principles,	46
55.	production of ions by electron impact, chemical ionization and field desorption	47-48
55.	techniques,	17 10
56.	separation and detection of ions	49-50
57.	mass spectrum: molecular ion peak, base peak, isotopic peak, metastable peak;	51
58.	fragmentation patterns of organic molecules with examples of various classes of	52-53
00.	compounds,	0200
59.	McLafferty rearrangement,	54
60.	factors affecting the fragmentation and governing the reaction pathways,	0.1
61.	identification of molecular ion peaks,	55
62.	determination of molecular weight and molecular formula of compounds,	
63.	hydrogen deficiency index, nitrogen rule,	56-57
<u>64</u> .	negative ion mass spectrometry,	2021
65.	introduction to high resolution mass spectrometry (HRMS) and combined or	58-59
	hyphenated techniques likes GC-MS, LC-MS, IC-MS, CE-MS, ICP-MS;	20.07
66.	applications mass spectrometry	60
	V: Structure Elucidation:	
67.	An integrated problem solving approach based on analytical data including	61-75
	CHNS/O percentage, spectral data (UV, IR, NMR, MS, <i>etc.</i> ) and hyphenated	
	technique data (GC-MS, LC-MS, ICP-MS, LC-NMR, <i>etc.</i> ) including reaction	
	sequences for structure elucidation of organic compounds	

- Encyclopedia of Spectroscopy and Spectrometry, Three-Volume Set: Encyclopedia of Spectroscopy and Spectrometry, Second Edition: 3 volume set
- NMR Spectroscopy: Basic Principles, Concepts, and Applications in Chemistry, Harald Günther, Wiley; 2 edition, 1995.

- Carbon-13 NMR spectroscopy, Hans-Otto Kalinowski, Stefan Berger, Siegmar Braun, Wiley, 1988.
- Introduction to Spectroscopy, Donald L. Pavia, Cengage Learning, 2009
- Pulse methods in 1D and 2D liquid-phase NMR Wallace S. Brey, Academic Press, 1988.
- Organic Structure Determination Using 2-D NMR Spectroscopy: A Problem-Based Approach, Jeffrey H. Simpson, Academic Press, 2008.
- High-Resolution NMR Techniques in Organic Chemistry, Timothy D. W. Claridge, Elsevier, 1999
- Identification of Organic Compounds, R. M. Silverstien, G. C. Hassler and T. C. Morill, John Wiley.
- Organic Spectroscopy, Jag Mohan, Narosa Publication.
- Spectroscopy of Organic Compounds, P. S. Kalsi, New Age International.
- NMR, NQR, EPR and Mossbauer Spectroscopy in Inorganic Chemistry, R. V. Parish, Ellis Harwood.
- Physical Methods in Chemistry, R. S. Drago, Saunders College.
- Introduction to Magnetic Resonance, A. Carrington and A. D. Maclachalan, Harper & Row.
- Gas Chromatography and Mass Spectrometry: A Practical Guide, Second Edition by O. David Sparkman
- Instrumental Methods of Chemical Analysis, Gurdeep Raj Chatwal and Shaym Anand, Himalaya Publications

# **Only for Organic Chemistry Specialization (Paper 3.3 & 3.4)**

### Paper-3.3: CHEM-633: Organic Synthesis

S.	Unit wise Broad Topics of the Syllabus	Tentative	
No.		Lecture(s) Allotted	
	Unit-I: Disconnection Approach-I:		
1.	Introduction, synthons and synthetic equivalents,	1	
2.	functional group inter-conversions,	2	
3.	order of events,	3	
4.	one and two group C-X disconnections,	4	
5.	chemoselectivity,	5-6	
6.	reversal of polarity,	7-8	
7.	cyclization reactions,	9-10	
8.	amine synthesis.	11	
9.	<b>Protecting Groups:</b> Principle of protection of alcohol, amine, carbonyl and carboxyl groups,	12-14	
10.	simple practices / exercises	15	
Unit-	II: Disconnection Approach-II:		
11.	One group C-C-disconnections involving alcohols and carbonyl compounds,	16-17	
12.	stereoselectivity,	18-19	
13.	regioselectivity,	20-21	
14.	alkene synthesis,	22	
15.	use of acetylenes.	23	
	Two group C-C disconnections in Diels-Alder reactions, 1,3-difunctionalised compounds and $\alpha$ - $\beta$ -unsaturated carbonyl compounds,	24-26	
16.	control in carbonyl condensations,	27	
17.	1,5-difunctionalised compounds,	28	
18.	Michael addition and Robinson annelation	29-30	
Unit-	III: Oxidation:		
19.	Introduction, different oxidative processes,	31	
20.	hydrocarbons: alkenes, saturated C-H groups (activated and inactivated), aromatic rings;	32-35	
21.	alcohols and diols;	36-37	
22.	aldehydes and ketones, ketals,	38-40	
23.	carboxylic acids,	41-42	
24.	amines, hydrazines and sulfides;	43	

25.	oxidations with ruthenium tetraoxide, iodobenzene diacetate and thallium (III)	44-45
	nitrate	
Unit	-IV: Reduction:	
26.	Introduction, different reductive processes,	46
27.	hydrocarbons: alkenes, alkynes and aromatic rings;	47-50
28.	carbonyl compounds: aldehydes, ketones;	51-54
29.	acids and their derivatives;	55-56
30.	epoxides;	57
31.	nitro, nitroso, azo and oxime groups;	58-59
32.	hydrogenolysis	60
	-V: Molecular Rearrangements:	
33.	General mechanistic considerations,	61
34.	nature of migration,	
35.	migratory aptitude,	62
36.	memory effects,	
37.	Pinacol-pinacolone rearrangement,	63
38.	Wagner-Meerwein rearrangement,	64
39.	Tiffeneu-Demjanov rearrangement,	65
40.	Dienone-Phenol rearrangement,	66
41.	Wolff rearrangement;	67
42.	Beckmann rearrangement,	68
43.	Hoffman rearrangement,	69
44.	Curtius rearrangement,	
45.	Lossen rearrangement,	70
46.	Schmidt rearrangement,	
47.	Baeyer-Villiger rearrangement,	71
48.	Benzil-Benzilic acid rearrangement,	72
49.	Favorskii rearrangement,	73
50.	Neber rearrangement,	
51.	Wittig rearrangement,	74
52.	Fries rearrangement,	
53.	Benzidine rearrangement	75

- Organic Synthesis: The Disconnection Approach, Stuart Warren, John Wiley & Sons.
- Organic Synthesis through Disconnection Approach, P. S. Kalsi
- Organic Synthesis, Smith M. B. McGraw Hill
- Modern Organic Synthesis, G. S. Zweifel and M. H. Nantz, Freeman and Company, New York.
- Modern Synthetic Reactions. H.O. House, W.A. Benjamin.
- Some Modern Methods of Organic Synthesis, W. Carruthers, Cambridge Univ. Press.
- Structure and Mechanism in Organic Chemistry, C.K. Ingold, Cornell University Press.
- Organic Chemistry, Clayden, Nick Geeves and Staurt Warren, Oxford University Press Principles of Organic Synthesis, R.O.C. Norman and J.M. Coxon, Blackie Academic & Professiona.
- Advanced Organic Chemistry, Reactions Mechanisms and Structure, J. March. John Wiley.
- Advanced Organic Chemistry Part B.F.A. Carey and R.J. Sundberg Plenum Press.
- Rodd's Chemistry of Carbon Compounds. Ed. S. Coffey, Elsevier

S.	Unit wise Broad Topics of the Syllabus	Tentative
No.		Lecture(s) Allotted
	I: Nomenclature of Heterocycles:	1
1.	Trivial, systematic (Hantzsch-Widman system), fusion, replacement systems of	1-3
	nomenclature for monocyclic, fused, spiro and bridged heterocycles	
	Aromatic Heterocycles:	
2.	General chemical behavior,	4
3.	classification (structural type),	
4.	aromaticity in heterocycles: relationship with carbocyclic aromatic compounds,	5
5.	criteria of aromaticity (structural, electronic, energetic and magnetic criteria);	6-7
6.	heteroaromatic ring systems,	8
7.	heteroaromatic reactivity and tautomerism in aromatic heterocycles	
	Non-aromatic Heterocycles:	
8.	Strain, bond angle strain and torsional strain and their consequences of in small	9-10
	ring heterocycles,	
9.	conformation of flexible heterocycles: five-membered & six-membered	11-12
	heterocycles	
10.	stereo-electronic effects in saturated six-membered heterocycles: anomeric and	13-14
	related effects;	
11.	attractive interactions through space (hydrogen bonding and nucleophilic-	15
	electrophilic interactions)	
Unit-	II: Three- and Four-membered Heterocycles :	
12.	Three-membered heterocycles with one heteroatom: Syntheses and reactions of	16-20
	aziridines, azirines, oxiranes, thiiranes.	
13.	Three-membered heterocycles with two heteroatom: Syntheses and reactions of	21-24
	diaziridines, diazirines, oxaziridines.	
14.	Four-membered heterocycles with one heteroatom: Syntheses and reactions of	25-30
	azetidines, azetidinones, oxetanes, oxetanones, thietanes	
Unit-	III: Five-membered Heterocycles:	L
15.	Five-membered heterocycles with one heteroatom: structure, stability, basicity,	31-33
	aromaticity, reactivity, synthesis and reactions of pyrrole, furan, thiophene.	
16.	Benzo-fused five-membered heterocycles with one heteroatom: synthesis,	34-37
	reactions and some medicinal importance of benzopyrroles.	
17.	Five-membered heterocycles with two heteroatoms: structure, reactivity,	37-40
	synthesis, reactions and some medicinal importance of 1,2-diazoles (pyrazoles	0, 10
	and isoxazoles) and 1,3-diazoles (imidazoles and oxazoles).	
18.	Benzo-fused five-membered heterocycles with two heteroatoms: synthesis,	42-43
10.	reactions and some medicinal importance of benzimidazoles.	12 13
19.	Five-membered heterocycles with more than two heteroatoms: synthesis,	44-45
17.	reactions and some medicinal importance of tetrazoles.	11 15
I Init.	IV: Six-membered Heterocycles-I:	
20.	Six-membered heterocycles with one and two nitrogen heteroatom: synthesis,	46-52
20.	reactions and some medicinal importance of azines (pyridines), diazines	+0-52
	(pyradizine, pyrimidine and pyrazine) and triazines (s-triazines).	
21	Benzo-fused six-membered heterocycles with one and two nitrogen heteroatom:	53-60
21.		55-00
	synthesis, reactions and some medicinal importance of quinoline, isoquinoline, quinazoline and quinovaline	
I In:4	quinazoline and quinoxaline.	
	V: Six-membered Heterocycles-II:	61.62
22.	Six-membered heterocycles with one oxygen heteroatom: synthesis and	61-63

## Paper-3.4: CHEM-634: Heterocyclic Chemistry

	reactions of pyrylium salts and pyrones.	
23.	Benzo-fused six-membered heterocycles with one oxygen heteroatom: synthesis,	64-67
	reactions and some medicinal importance of coumarins and chromones.	
	Seven-membered Heterocycles	
24.	Synthesis and some medicinal importance of benzodiazepines, benzoxazepines	68-73
	and benzothiazepines.	
	Large-membered Heterocycles:	
25.	Synthesis and some medicinal importance of azocines and oxocines	74-75

### Books:

- Heterocyclic Chemistry Vol. 1-3, R.R. Gupta, M. Kumar and V. Gupta, Springer Verlag.
- The Chemistry of Heterocycles, T. Eicher and S. Hauptmann, Thieme.
- Heterocyclic Chemistry, J.A. Joule, K. Mills and G.F. Smith, Blackhall.
- Heterocyclic Chemistry, T.L. Gilchrist, Longman Scientific Technical.
- Contemporary Heterocyclic Chemistry, G,.R. Newkome and W.W. Paudler, Wiley-Inter Science.
- An Introduction to the Heterocyclic Compounds, R.M. Acheson, John Wiley.
- Comprehensive Heterocyclic Chemistry, A.R. Katrizky and C.W. Rees, eds. Pergamon Press

# **Only for Analytical Chemistry Specialization (Paper 3.3 & 3.4)**

## Paper-3.3: CHEM-633: Advanced Analytical Techniques

S.	Unit wise Broad Topics of the Syllabus	Tentative	
No.		Lecture(s) Allotted	
Unit-	Unit-I: Extraction Methods:		
1.	Basic principles, classification of extraction systems,	1	
2.	factors affecting extraction process,	2-3	
3.	mechanism of extraction,	4	
4.	extraction of liquids,	4	
5.	extraction by chelation, extraction by solvation, extraction by ion-pair formation, extraction by sonication	5-7	
6.	extraction equilibria for chelates, extraction equilibria for solvation,	8-9	
7.	separation of metals by extraction,	10	
8.	solid-phase extraction (SPE),	11	
9.	supercritical fluid extraction (SFE),	12	
10.	supramolecular extraction,	13	
11.	centrifugation and ultra-centrifugation,	14	
12.	membrane separation,	15	
Unit-	II: Isolation and Purification Techniques:		
13.	Filtration (simple, micro, etc.),	16-17	
14.	recrystallization (in aqueous & non-aqueous solutions, at low temperature, in inert atmosphere, semi-micro & micro),	18-20	
15.	use of decolourising carbon,	21-22	
16.	difficulties in recrystallization,		
17.	drying of liquids,	23	
18.	freezing, sublimation,		
19.	distillation (simple, steam, fractional, vacuum, high vacuum or molecular, etc.),	24-25	
20.	nucleation and crystal growth,	26	
21.	crystal hydrates and solvates,	27	
22.	chemical methods for separation,	28	

23.	determination of physical constants (mp, mixed mp, bp, m. wt., density, optical	29-30
	rotator power, RI, etc.)	
Unit	-III: High Frequency Titrations:	
24.	Principle, Instrumentation-Cells, oscillator circuit and high frequency titrimeters,	31-34
25.	theory,	35
26.	correlation of high frequency titration curves with low frequency titration	36-38
	curves.	
27.	Applications-acid base, complexometric, measurement of dielectric constant and	39-43
	analysis of mixture of organic compounds	
28.	Advantages and disadvantages of high frequency methods	44-45
Unit	-IV: Polarography:	
29.	Principles, classification of polarographic techniques,	46-47
30.	types of polarographic currents,	48
31.	instrumentation,	49-50
32.	factors affecting polarographic wave, pulse polarography, and differential pulse	51-52
	polarograph.	
	Potentiometry:	
33.	Metal electrodes for measuring the metal's cation, metal-metal salt electrodes,	53-54
	redox electrodes, calomel electrode,	
34.	measurement of potential,	55
35.	determination of concentrations, residual liquid-junction potential,	56-57
36.	accuracy on direct potentiometric,	58
37.	glass pH electrode,	59
38.	ion-selective electrodes	60
Unit	-V: Voltammetry:	
39.	Voltammetric principles,	61
40.	hydrodynamic voltammetry, stripping voltammetry, cyclic voltammetry,	62-65
41.	criteria of reversibility of electrochemical reactions,	66-67
42.	quasi-reversible and irreversible processes,	68-69
43.	qualitative and quantitative analysis	70
	Amperometry	
44.	Principles and amperometric titration techniques: Dropping mercury electrode,	71-75
	rotating platinum microelectrode and dead stop	

- Introduction to Instrumental Analysis, R. D. Braun, McGraw-Hill Book Company, New Delhi
- Vogel's Textbook of Quantitative Chemical Analysis, 6th Edn. Pearson Education Asia
- Analytical Chemistry, An Introduction, D. A. Skoog, D. M. West, F. J. Holler, and S. R. Crouch, 7th. Edn., Saunders College publishing, N. Y.
- Principles of Instrumental Analysis, D. A. Skoog and J. J. Leary, 4th Edn., N. Y.
- The Principles of Electrochemistry, A. Duncan, Mac Innes Dover Publication Inc. N. Y.F. Scholz, Electroanalytical methods, Springer, 2002.
- P. Monk, Fundamentals of electroanalytical chemistry, Wiley, 2001.
- A.P.F. Turner I. Karube, I. G. Wilson, Biosensors- Fundamentals and applications. Oxford University Press, New York, 1987.
- Organic electro chemistry by Henning Lund & Ole Hammerich, , 4th edition, Publisher: Marcel Dekker, Inc, New York.

S.	Unit wise Broad Topics of the Syllabus	Tentative
No.	L. Analysis of Detrochomicals:	Lecture(s) Allotted
	-I: Analysis of Petrochemicals:	1
1.	Constituents, petroleum fractionation,	1 2
2	analysis of petroleum products:	3-4
2. 3.	specific gravity, viscosity, doctor test, sulphuric acid absorption,	5-6
<u> </u>	aniline point, cloud point, pour point; determination of water,	7
<del>4</del> . 5.	neutralization value (acid and base numbers),	8
<u> </u>	ash content, sulphur and mercapton sulphur,	9-10
7.	determination of lead in petroleum	11
8.	vapour pressure and colour determination,	11
<u>9.</u>	Analysis of Fuels: Proximate and ultimate analysis of fuel,	13
<u>9.</u> 10.	calorific value by Bomb calorimetry,	13
11.	analysis of fuel gases (coal gas, producer gas, water gas)	15
	-II: Analysis of Agrochemicals:	15
Umu	Analysis of Fertilizers:	
12.	Analysis of Tertifizers. Analysis of nitrogen: urea nitrogen, total Kjeldahl nitrogen method, ammonia	16-18
12.	nitrogen;	10-10
13.	analysis of phosphrous: total phosphrous, available and non-available,	19-21
15.	alkalimetric ammonium molybdophosphate method;	17-21
14.	analysis of potassium: potassium by sodium tetraphenyl borate method.	22
15.	Analysis of potassiani potassiani by sodiani etraphenyi bolate method. Analysis of herbicides: atrazine, alachlor; Analysis of Fungicides: nimbin,	23-24
15.	carbendazim;	25-24
16.	Analysis of bactericides: chloramine, triclosan, chlorhexidine;	25
17.	Analysis of insecticides: DDT, BHC, aldrin, endosulfan, malathion,	26-28
17.	monochrotophos;	20 20
18.	Analysis of nematicides: aldicarb; Analysis of rodenticides: warfarin,	29-30
101	bromadiolone	
Unit	-III: Analysis of Polymers:	I
19.	Chemical Methods of Analysis: Introduction, preparation of the sample,	31-35
	determination of purity, physical tests, preliminary examination, burning	
	characteristics, transition points,	
20.	molecular weight, density, refractive index,	36-37
21.	pyrolytic behaviour,	38-40
22.	qualitative and quantitative elementary analysis,	
23.	solubility and acid numbers, acetyl number, iodine number	41
24.	end group analysis, colour tests	42
25.	Analysis of Plastics	
26.	Basics of plastic analysis, fundamental conditions for plastic analysis,	43
27.	water test, copper wire test, acetone test, heat test, isopropyl alcohol test, oil test	44-45
Unit	IV: Analysis of Glass and Ceramics:	
28.	Introduction, composition, method of analysis-sampling and sampling	46-47
	preparation,	
29.	composition analysis-preliminary testing, decomposition, chemical method for	48-49
	the individual constituents-Si, B, Pb, Zn, Al, Cl, Ca, Mg, Ti	
30.	Analysis of Cement:	
31.	Loss on ignition, insoluble residue,	46-50

## Paper-3.4: CHEM-634: Analysis of Commercial Products

32.	total silica, sesquioxiodes, lime,	51-53
33.	magnesia, ferric oxide, sulphuric anhydride, air and dust pollution from cement	54-60
	plants, atmospheric dispersion of pollutants in cement industry	
Unit	-V: Analysis of Minerals and Ores:	
34.	Hematite,	61-62
35.	pyrolusite,	63-64
36.	gypsum,	65
37.	dolomite,	66
38.	bauxite,	67
39.	Chromate	68
40.	limestone and	69
41.	uranium ores	70
	Analysis of Metal and Alloys:	
42.	Steel, Cu-Ni alloy,	71
43.	solder, bronze, brass,	72
44.	aluminium alloy, ferroalloys of silicon, chromium, titanium and vanadium	73-75

- Introduction to Instrumental Analysis, R. D. Braun, McGraw-Hill Book Company, New Delhi
- Vogel's Textbook of Quantitative Chemical Analysis, 6th Edn. Pearson Education Asia
- Analytical Chemistry, An Introduction, D. A. Skoog, D. M. West, F. J. Holler, and S. R. Crouch, 7th. Edn., Saunders College publishing, N. Y.
- Principles of Instrumental Analysis, D. A. Skoog and J. J. Leary, 4th Edn., N. Y.
- The Principles of Electrochemistry, A. Duncan, Mac Innes Dover Publication Inc. N. Y.F. Scholz, Electroanalytical methods, Springer, 2002.
- P. Monk, Fundamentals of electroanalytical chemistry, Wiley, 2001.
- A.P.F. Turner I. Karube, I. G. Wilson, Biosensors- Fundamentals and applications. Oxford University Press, New York, 1987.
- Organic electro chemistry by Henning Lund & Ole Hammerich, , 4th edition, Publisher: Marcel Dekker, Inc, New York.



# **Department of Pure and Applied Chemistry** University of Kota, Kota

# M.Sc. IV Semester Chemistry

# Lecture Plans for Academic Session 2018-19

# **<u>Common Papers for All Specializations</u>**

# Paper-4.1: CHEM-641: Environmental Chemistry

S. No.	Unit wise Broad Topics of the Syllabus	Tentative Lecture(s) Allotted
Unit-	I: Air Pollution:	
1.	Concept of environment chemistry,	1
2.	composition of atmosphere,	
3.	major sources of air pollution, chemical reactions, smog formation, acid rain,	2-3
4.	classification and effect of air pollutants, NOx, SOx, COx particulates and ozone;	4-5
5.	Greenhouse effect and global warming,	6
6.	ozone depletion,	7
7.	automobile emissions,	
8.	prevention and control of vehicular pollution,	8
9.	alternative fuels: Biodiesel, ethanol, CNG, ultra low sulphur diesel (ULSD)	9
	Monitoring of Air Pollution:	
10.	Principles of environment monitoring,	10
11.	methods for monitoring of air pollutants including NOx, SOx, COx, SPM	11
	Prevention and Control of Air Pollution:	
12.	Control of pollution by fuel selection and utilization,	12
13.	process or equipment modification,	13
14.	devices, site selection, stacks, planting trees and growing vegetation,	14
15.	general methods of air pollution control	15
Unit-	II: Water Pollution:	
16.	Types of water pollution,	16
17.	sources of water pollution,	17
18.	water pollutants, their classification and effects,	18-19
19.	water pollution laws and standards	20
	Analysis of Water	
20.	Chemical and physical examination of water,	21
21.	preservation and pre-concentration,	22
22.	hydrogen ion concentration, acidity, alkalinity, hardness, pH, free CO <sub>2</sub> , Cl <sub>2</sub> , metals, ions, dissolved chlorine and oxygen, BOD, COD, chlorine dosage, <i>E. coli</i> index,	23-29
23.	general methods of water pollution control	30
Unit-	III: Soil Pollution :	
24.	Composition and types of soil,	31
25.	mineral and organic matter in soil,	

26.	soil pollution by industrial wastes, urban wastes, radioactive pollution and	32-34
20.	agriculture practices	52 51
	Soil Analysis:	
27.	Analysis of nitrates, nitrites, ammonical nitrogen, total nitrogen, phosphates,	35-40
	organic carbon, potassium, calcium, sodium, magnesium, iron, zinc, etc.	
	Control of Soil Pollution:	
28.	Control of domestic and industrial wastes,	41-43
29.	soil remediation,	44
30.	environmental friendly technologies for agriculture	45
Unit	-IV: Industrial Pollution:	
31.	Environmental pollution from various industries and control of industrial	46-47
	pollution	
	Industrial Wastes and their Treatment:	
32.	Characteristics and types of industrial wastes,	48
33.	principles of industrial waste treatment,	49-50
34.	protection of biosphere and surface water from pollution with industrial	51-53
	sewages,	
35.	sampling and chemical analysis of industrial waste water,	54-56
36.	waste water treatment,	57
37.	solid waste management,	58
38.	hazardous waste management	59-60
Unit	-V: Radioactive Pollution:	
39.	Radioactive substances,	61
40.	state of radioactive isotopes in solution, gases and solids;	62-64
41.	units of radiation,	65
42.	analysis of radionuclides,	66
43.	sources of radioactive pollution,	67
44.	radioactive fallout,	68
45.	nuclear reactors,	69
46.	nuclear installations,	70
47.	radioactive ore processing,	71
48.	nuclear accidents,	72
49.	effects of radioactive pollution on power plants and polymers,	73-74
50.	control of radioactive pollution	75

Books:

• Environmental Chemistry. B. K. Sharma. 12<sup>th</sup> Edition, 2011, Goel Publishing House, Meerut.

• Environmental Chemistry, Colin Baird, W.H. Freeman Co. New York, 1998.

• Environmental Pollution: Principles, Analysis and Control. P. Narayanan. 1<sup>st</sup> Edition, 2007, CBS Publishers.

• Environmental Pollution Control Engineering. C. S. Rao. 2<sup>nd</sup> Edition, 2006, New Age International Publishers.

• Environmental Pollution analysis, S.M. Khopkar, Wiley Eastern, New Delhi, 1994.

• Pollution Control in Process Industries. S. P. Mahajan. 20th Ed, 2006, TataMcGraw-Hill, New Delhi.

• Industrial Pollution. V. P. Kudesia. 5<sup>th</sup> Edition, 2007, Pragati Prakashan, Meerut.

• Water Supply and Sanitary Engineering. G. S. Birdie & J. S. Birdie. 8<sup>th</sup> Edition, 2008, Dhanpat Rai Publication.

• Environmental Toxicology, J.Rose Gordon and Breach (Ed.), Science Publication, New York, 1993.

• Introduction to Atmospheric Chemistry, P.V. Hobbs, Cambridge.

S. No.	Unit wise Broad Topics of the Syllabus	Tentative Lecture(s) Allotted
Unit-	I: Modern Approaches of Organic Synthesis:	
1.	Principles and concepts of green chemistry,	1
2.	atom economy,	2
3.	waste minimization techniques,	3
4.	different approaches to green synthesis	4
	Reagents:	
5.	Dimethyl carbonate;	5
6.	polymer supported reagents: chromic acid and per-acids	6
	Catalysts:	
7.	Introduction to catalysts,	7
8.	homogeneous and heterogeneous catalysts,	8
9.	solid acid-base catalysts,	9
10.	basic catalysts,	10
11.	metal oxide supported catalysts,	11
12.	oxidation catalysts,	12
13.	polymer supported catalysts,	13
14.	phase transfer catalysts,	14
15.	bio-catalysts	15
Unit-	II: Solvents for Organic Synthesis:	
16.	Introduction, characteristics properties, types and examples of green solvents	16-17
	Water:	
17.	Reasons for using water as green solvent,	18
18.	biphasic systems,	
19.	synthesis in water (asymmetric aldol reaction, synthesis of quinoxalines, carbon	19-21
	dioxide fixation, preparation of nanoparticles),	
20.	near critical water	22
	Supercritical Liquids:	
21.	The phase diagram of $CO_2$ ,	23
22.	supercritical CO <sub>2</sub> ,	
23.	properties and applications of $CO_2$ in dry cleaning, decaffeination of coffee and	24
	synthesis	
	Ionic Liquids:	
24.	Basic concept, types, physicochemical properties,	25
25.	preparation of ionic liquids: dialkylimidazolium and alkylpyridinium cation	26-27
	based ionic liquids, ionic liquids with fluorine containing anions and chiral ionic	
	liquids;	
26.	synthetic applications of ionic liquids (alkylation, allylation, oxidation and	28-29
	hydrogenation),	
27.	concept of supported ionic liquids and their applications	30
Unit-	III: Microwave Assisted Organic Synthesis:	
28.	Introduction of microwave assisted organic syntheses,	31
29.	fundamentals of microwave technology,	32-33
30.	microwave activation,	34-35
31.	equipment,	36
32.	time and energy benefits,	37-38
33.	limitations;	

# Paper-4.2: CHEM-642: Recent Methods of Organic Synthesis

34.	applications,	
35.	reactions in organic solvents: Esterification, Diels-Alder reaction,	39-45
	decarboxylation; solvent free reactions (solid state reactions): Deacetylation,	
	deprotection, saponification, alkylation of reactive methylene compounds	
Unit-	IV: Ultrasound Assisted Organic Synthesis:	
36.	Basics of sono-chemistry,	46
37.	ultrasound cavitation,	47-48
38.	sonocemical effect,	49-50
39.	experimental parameters,	51-52
40.	transducers,	53
41.	reactors,	54-55
42.	homogeneous and heterogeneous sono-chemistry,	56-57
43.	oxidation, reduction, substitution reactions, Kornblum-Russell reaction, Hetero	58-60
	Micahel reaction, preparation of Grignard's reagent	
Unit-	V: Organic Synthesis Using Reactors:	
44.	General introduction and types of reactors,	61
45.	chemical reactor design,	62
46.	simulation and optimization;	63
47.	mass and energy balance,	64-65
48.	mass and energy transfer.	66-67
49.	Batch reactors: Basic concepts, types and reactions;	68
50.	concepts of laboratory and pilot scale organic syntheses.	69
51.	Vapour phase reactors: Types and design. Raw materials, process flow diagrams,	70-72
52.	product syntheses, separations, purifications and waste compositions at	73-75
	industrial scale productions of pharmaceuticals, agrochemicals, organic	
	fertilizers and dyes	

### Books:

- Green Chemistry: Theory and Practice, Paul T. Anastas and John C. Warner
- Green Chemistry: An Introductory Text by Mike Lancaster, Royal Society of Chemistry
- Green Chemistry and Catalysis by Sheldon, Arends and Hanefeld, WILEY-VCH, Germany
- Green Solvents, Vol. 5: Reactions in Water. edited by Paul T. Anastas, WILEY-VCH
- Green Solvents, Vol. 6: Ionic Liquids. edited by Paul T. Anastas, WILEY-VCH
- Ionic Liquids in Synthesis by Wasserscheid and Welton. WILEY-VCH
- Microwaves in Organic Synthesis, Antonio de la Hoz (Ed), André Loupy (Ed), Wiley-VCH
- Organic Synthesis in Water, Paul A Grieco Blackie.
- Organic Synthesis: Special Techniques, V. K. Ahluwalia and Renu Aggrawal
- Chemical Reviews 2007, 107, 2167-2820 (Special issue on Green Chemistry)
- Fundamentals and Applications of Organic Electrochemistry: Synthesis, Materials, Devices by Toshio Fuchigami, Mahito Atobe, Shinsuke Inagi.

# **Only for Organic Chemistry Specialization (Paper 4.3 & 4.4)**

## Paper-4.3: CHEM-643: Chemistry of Natural Products

<b>S.</b>	Unit wise Broad Topics of the Syllabus	Tentative
No.		Lecture(s) Allotted
Unit-	I: Terpenoids and Carotenoids:	
1.	Classification, nomenclature, occurrence, isolation,	1
2.	general methods of structure determination,	2
3.	isoprene rule,	
	stereochemistry and synthesis of the following representative molecules:	

4.	citral,	3-4
5.	geraniol,	5
6.	α-terpineol,	6-7
7.	menthol,	8
8.	farnesol,	9
9.	zingiberene,	10-11
10.	abietic acid,	12-13
11.	β-carotene	14-15
	II: Alkaloids:	
12.	Definition, nomenclature and physiological action, occurrence, isolation,	16
13.	identification (qualitative idea only),	17
14.	general methods of structure elucidation,	18-19
15.	degradation,	20
16.	classification based on nitrogen heterocyclic ring,	21
17.	role of alkaloids in plants;	22
17.	structure, stereochemistry and synthesis of the following:	
18.	ephedrine,	23-24
19.	coniine,	25 25
20.	nicotine,	26
21.	atropine,	20
22.	quinine,	28
23.	morphine	29-30
	III: Steroids and Hormones :	27-30
24.	Occurrence, nomenclature, basic skeleton,	31
25.	Diel's hydrocarbon and stereochemistry,	32
26.	isolation, identification (qualitative idea only),	32
20.	structure determination and synthesis of following:	55
27.	cholesterol,	34-35
28.	bile acids,	36
29.	androsterone,	37-38
30.	testosterone,	39-40
31.	oestrone,	41-42
32.		41-42
33.	progesterone, aldosterone	45
		45
	IV: Prophyrins and Plant Pigments: Structure and synthesis of haemoglobin and chlorophyll	46-48
34.		40-48
	Plant Pigments:	40.50
	Occurrence, nomenclature, isolation, general methods of structure determination,	49-50
25	and synthesis of following	51
35.	apigenin,	51
36.	luteolin,	52
37.	quercetin,	53
38.	myrcetin,	54
39.	diadzein,	55
40.	cyanidin,	56
41.	cyanidin-7-arabinoside,	57
42.	Hirsutidin.	58
12	Biosynthesis of Flavanoids:	<b>50</b> (0
43.	Acetate pathway and shikimic acid pathway.	59-60

Unit	Unit-V: Prostaglandins, Pyrethroids and Rotenones:		
44.	Occurrence, nomenclature, classification, biogenesis and physiological effects,	61-62	
45.	synthesis of PGE <sub>2</sub> and PGF <sub>2<math>\alpha</math></sub>	63-68	
	Pyrethroids and Rotenones:		
46.	Synthesis and reactions of pyrethroids and rotenones (for structure elucidation,	69-75	
	emphasis is to be placed on the use of spectral parameters wherever possible)		

Books:

- Natural Products : Chemistry and Biological Significance, J. Mann, R.S. Davidson, J.B. Hobbs, D.V. Banthrope adn J.B. Harbome, Longman, Essex.
- Organic Chemistry: Vol. 2, I.L. Finar, ELBS.
- Stereoselective Synthesis: A Practical Approach, M. Norgradi, VCH.
- Chemistry, Biological and Pharmacological Properties of Medicinal Plants from the Americas, Ed. Kurt Hostettmann, M.P. Gupta and A. Marston. Harwood Academic Publishers.
- Introduction to Flavonoids, B.A. Bohm. Harwood Academic Publishers.
- New Trends in Natural Product Chemistry, Ata-ur-Rahman & M.L. Choudhary, Harwood Academic Publishers

S. No.	Unit wise Broad Topics of the Syllabus	Tentative Lecture(s) Allotted
Unit-	I: Drug Design:	
1.	Procedures followed in drug design & development,	1
2.	concepts of lead compound and lead modification, pro drugs and soft drugs, isosterism, non-isosterism,	2
3.	factors affecting bioactivity,	3
4.	theories of drug activity: occupancy theory, rate theory, induced fit theory;	4
5.	computer-aided drug design,	5
6.	quantitative structure activity relationship,	6
7.	concepts of drug receptors, elementary treatment of drug receptor interactions;	7
8.	physico-chemical parameters: lipophilicity, partition coefficient, electronic ionization constants, steric,	8
9.	Shelton and surface activity parameters and redox potentials;	-
10.	Free-Wilson analysis,	9
11.	Hansch analysis,	
12.	relationships between Free-Wilson and Hansch analysis,	
13.	LD-50 and ED-50 (Mathematical derivations of equations excluded)	10
	Pharmacokinetics & Pharmacodynamics:	
	Pharmacokinetics:	
14.	Introduction to drug absorption, disposition, elimination using pharmacokinetics;	11-12
	Pharmacodynamics:	
15.	Enzyme stimulation, enzyme inhibition,	13
16.	membrane active drugs,	
17.	drug metabolism, factors affecting drug metabolism,	14
18.	biotransformation,	15
19.	significance of drug metabolism in medicinal chemistry	
	II: Anti-cancer Drugs:	
20.	Introduction, cellular apoptosis,	16-17
21.	oncogenes, tumor suppressor genes,	18
22.	disease states, cancer chemotherapy.	19

### Paper-4.4: CHEM-644: Medicinal Chemistry

23.	Mechanism of action and structure activity relationship.	20-21
	Synthetic procedures of	
24.	mechlorethamine, cyclophosphamide,	22
25.	capacetabine, 6-mercaptopurine, trimetrexate,	23
26.	dactinomycin, daunomycin,	24
27.	etoposide, irinotecan,	25
28.	paclitaxel, imatinib	26
	Anti-viral Drugs:	
29.	Classification of viruses,	27
30.	infectious process, mechanism of action,	
31.	structure activity relationship	28
	Synthetic procedures of	
32.	idoxuridine, zidovudine,	29
33.	nevirapine, efavirenz, squinavir	30
Unit	III: Cardiovascular Drugs :	
34.	Cardiovascular diseases,	31
35.	drug inhibitors of peripheral sympathetic function,	32
36.	central intervention of cardiovascular output,	33
37.	intermediatory myocardial metabolism,	34
38.	cardiac electrophysiology	35
39.	Mechanism of action and structure activity relationship	36
	Synthetic procedures of	
40.	nitroglycerin, aspirin,	37
41.	verapamil, diltiazem,	
42.	procainamide, lidocaine, sotalol	38
	Anti-hypertensive Drugs	
43.	Mechanism of action, classes of anti-hypertensive drugs,	39
	Synthetic procedures of	
44.	digitalis, methyldopa,	40
45.	tolazoline, phenoxybenzamine,	41
46.	doxazosin, propanolol, acebutamol, labetalol,	42-43
47.	captopril,	44
48.	losartan	45
Unit	IV: Drugs Affecting Central Nervous System :	
	Anxiolytics, Sedatives and Hypnotics:	
49.	GABA receptor modulators,	46
50.	mechanism of action and structure activity relationship of benzodiazepines and	47-48
	barbiturates	
	Synthetic procedures of	
51.	diazepam, alprazolam, zolpidem,	49-50
52.	ramelteon, buspirone,	51
53.	phenobarbital, butabarbital, pentobarbital	52-53
	Anti-psychotics:	
54.	Neurotransmitters, neurochemistry of mental diseases,	54
55.	mechanism of action,	55
56.	structure activity relationship	56
	Synthetic procedures of	
57.	chlorpromazine,	57
58.	thiothixene, haloperidol,	58

59.	clozapine, olanzapine, quetiapine	59-60
Unit	-V: Anti-infective Drugs :	
	Mechanism of action, structure activity relationship and synthetic procedures of drugs of following classes	
60.	Anti-fungal Drugs: Fluconazole, caspofungin and griseofulvin	61-62
61.	Anti-bacterial Drugs: Ciprofloxacin, nitrofurazone and sulfamethazine	63-64
62.	Anti-tubercular Drugs: Isoniazid and rifampin	65
63.	Anti-protozoal Drugs: Metronidazole	66
64.	Anthelmintic Drugs: Albendazole	67
65.	Anti-malarial Drugs: Quinine, chloroquine, primaquine and artemisinin	68-69
	Analgesic Drugs:	
66.	Origin and types of pain,	70
67.	mechanism of action and structure activity relationship of steroidal and non- steroidal drugs.	71
	Synthetic procedures of	
68.	pentazocine, paracetamol, aspirin,	72-73
69.	ibuprofen, diclofenac, sumatriptan	74-75

### Books:

- Burger's Medicinal Chemistry and Drug Discovery All Volumes, Wiley.
- Wilson Gisvold's Text book of Organic Medicinal and pharmaceutical Chemistry, Ed. Robert F. Dorge.
- Foye's Principles of Medicinal Chemistry, David A. Williams, LWW.
- Introduction to Medicinal Chemistry, A Gringuage, Wiley- VCH.
- An Introduction to Medicinal Chemistry, Graham L. Patrick, Oxford University Press
- An Introduction to Drug Design, S. S. Pandeya and J. R. Dimmock, New age International.
- Goodman and Gilman's Pharmacological Basis of Therapeutics, Mc Graw-Hill.
- The Organic Chemistry of Drug Design and Drug Action, R.B. Silverman, Academic press.
- Strategies for Organic Drug Synthesis and Design, D. Lednicer, John Wiley.
- Handbook of Stability Testing in Pharmaceutical Development: Regulations, Methodologies, and Best Practices by Kim Huynh-Ba, Springer

# **Only for Analytical Chemistry Specialization (Paper 4.3 & 4.4)**

## Paper-4.3: CHEM-643: Instrumental Methods of Analysis

S.	Unit wise Broad Topics of the Syllabus	Tentative
No.		Lecture(s) Allotted
Unit	I: X-rays Diffraction :	
1.	Production of X-rays,	1
2.	X-rays spectra,	
3.	monochromatic X-rays sources,	2
4.	X-rays detectors,	
5.	X-rays absorption,	3
6.	X-ray fluorescence,	
7.	X-rays diffraction methods,	4
8.	Bragg's law,	
9.	determination of crystal structure by Bragg's law,	5-6
10.	XRD apparatus,	7-8
11.	applications of XRD in crystalite size determination by using Sherrer formula,	9-10
12.	determination of cis-trans isomerism, polymer crystallization	11

13.	X-ray photoelectron spectroscopy (XPS),	12
14.	X-ray fluorescence spectroscopy (XRF),	13
15.	Auger electron spectroscopy (AES),	14
16.	Energy-dispersive X-ray spectroscopy (EDS or EDX)	15
Unit	II: Electron Diffraction:	
17.	Scattering intensity v/s scattering angle,	16
18.	Wierl equation,	17
19.	measurement technique,	18-19
20.	elucidation of structure of simple gas phase molecules,	20-21
21.	low energy electron diffraction and structure of surfaces	22-23
	Neutron Diffraction:	
22.	Scattering of neutrons by solids and liquids,	24
23.	magnetic scattering,	25
24.	measurement techniques,	26-27
25.	elucidation of structure of magnetically ordered unit cell	28-29
26.	Microscopic Methods: General introduction to SEM and TEM	30
Unit	III: Thermo-analytical Methods:	
27.	Introduction and classification of thermoanalytical methods;	31
28.	thermogravimetric analysis (TGA): definition, types, instrumentation, TGA	32-35
	curve, factors affecting TGA curves, calculation of percent decomposition and	
	composition of compounds,	
29.	limitation and advantages of TGA, application of TGA to the thermal behavior	36-39
	including crystalline copper sulphate, calcium oxalate monohydrate, zinc	
	hexafluorosilicate;	
	differential thermal analysis (DTA): definition, theoretical basis,	40-42
	instrumentation, factors affecting the DTA curve,	
30.	application of DTA, advantages and disadvantages of DTA;	43
31.	differential scanning calorimetry (DSC): Definition, comparison of DTA and	44-45
	DSC techniques, instrumentation, factors affecting DSC curves	
	IV: Radio-analytical Methods:	
32.	Determination of nuclear radiation and counting devices,	46
33.	radioactivity tracers- principal and applications,	47-48
34.	isotopic analysis-direct and inverse,	49
35.	special analytical application-radiometric titrations,	50-51
36.	neutron activation analysis principle, instrumentation, applications and	52-53
	limitations,	
37.	radio-chromatography and radio-immunoassay	54-55
	Nephlometry and Turbidometry:	
38.	Introduction, theory, comparison of spectrophotometry, turbidimetry and	56-58
	nepelometry,	
39.	instrumentation and applications.	59-60
	V: Polarimetry:	
40.	Polarisation of light, optical activity, theories of optical activity,	61-62
41.	factors affecting angle of rotation,	63
42.	specific rotation,	
43.	optical rotator dispersion and circular dichroism-Cotton effect, ORD and CD	64
44.	curves, instrumentation, measurement of rotatory power, applications of polarimetry,	65-66
		67
45.	optical activity and chemical constitution,	07

47.	deciding between two structures for a molecule,	
48.	distinguish between a pair of enantiomorphs,	69
49.	saccharimetry, difference between saccharimetry and polarimetry, saccharimeters,	70-71
50.	kinetic polarimetry, spectropolarimetry	72
	Refractometry:	
51.	Principle, parameters influencing refraction,	73
52.	significance of critical angle during measurements,	
53.	refractometers,	74
54.	qualitative and quantitative analysis and analytical applications	75

Books:

- **Range of lectures required for completion of broad topics of the Syllabus : 65-75**
- D. A. Skoog and D. M. West, Fundamentals of Analytical Chemistry, Holt Rinehart and Winston Publications, IV Edn, 1982.
- D.A. Skoog, D.M. West, F.J. Holler and S.R. Crouch, Fundamentals of Analytical Chemistry, Thomson Asia Pte Ltd., Singapore, Viiith Edn., 2004.
- D.A. Skoog, Principles of Instrumental Analysis, Saunders College Pub.Co, III Edn., 1985.
- J.G. Dick, Analytical Chemistry, McGraw Hill Publishers, 1974.
- Willard, Merit, Dean and Settle, Instrumental Methods of Analysis, CBS Publishers and Distributors, IV Edn., 1989
- G. D. Christian and J.E.O Reilly, Instrumental Analysis, Allyn and Bacon Inc, II Edn., 1986.
- G.W. Ewing, Instrumental Methods of Chemical Analysis, McGraw Hill Pub, 1975

## Paper-4.4: CHEM-644: Analysis of Consumers Products

S. No.	Unit wise Broad Topics of the Syllabus	Tentative Lecture(s) Allotted		
Unit-I: Drug Analysis:				
1.	Introduction to drugs, their classification,	1		
2.	sources of impurities in pharmaceutical raw materials (chemical, atmospheric and microbial contaminants etc.);	2-5		
3.	impurity profile,	6-7		
4.	limit test, solubility tests, disintegration test, stability test,	8-10		
5.	narcotic and dangerous drugs,	11-12		
6.	analysis of some drugs (paracetamol, diclofenac, losartan, inidazole, alprazolam)	13-15		
Unit-	II: Clinical Analysis:			
7.	Sampling and selective analysis of biological fluids (using routine and automatic instruments): glucose, bilirubins & biluverdins, total cholesterol, haemoglobin, creatinine, total proteins, albumin, urea-nitrogen, carticosteroids and barbiturates;	16-15		
8.	vitamins and antibiotics;	26-27		
9.	immunological methods of analysis: ELISA and RIA	28-30		
Unit-	III: Food Analysis:			
10.	Sampling and selective analysis of food flavours, food colour, food preservatives, milk and milk products, floor starches, tea, coffee, sugar content analysis of honey, jam & jelly; alcohol content in beverages;	31-38		
11.	analysis of oils and fats: softening point, congent point, titre point, cloud point, iodine value, saponification value, acid value and Polenske value, Elaiden test;	39-44		
12.	pesticide residue analysis	45		
Unit-	IV: Cream & Lotion Analysis:			
13.	Composition of creams and lotions,	46		
14.	determination of water, propylene glycol, non-volatile matter and ash content,	47-50		

15.	determination of borates, carbonates, sulphates, phosphates, chlorides, titanium	51-54
	and zinc oxides	
	Face Powder, Deodorant & Antiperspirant Analysis	
16.	Composition,	55
17.	analysis of fats and fatty acids, boric acid, Mg, Ca, Zn, Fe, Ti, Al, phenol,	56-60
	hexachlorophenone, methanamine, sulphonates and urea	
Unit	-V: Soap Analysis:	
18.	Method of analysis: sampling, separation, identification;	61-62
19.	determination of fatty acids, total anhydrous soap and combined alkali,	63-68
	potassium, water, determination of inorganic fillers and soap builders;	
20.	determination of constituents and other additives	69
	Detergent Analysis	
21.	Types,	70
22.	method of analysis: sampling, separation, identification of components,	71-75
	determination of surfactants and other constituents	

#### Books:

- Standard Methods of Chemical Analysis, F. J. Welcher
- Instrumental Methods of Analysis (6th Edition) H. H. Willard & L. L. Merritt.
- A Text Book of Quantitative Inorganic Analysis (3rd Edition) A. I. Vogel.
- Treatise on Analytical Chemistry (Series of Volumes) I. M. Kolthoff & P. J. Elwing.
- Introduction to Instrumental Analysis R. D. Braun.
- Handbook of Industrial Chemistry Davis Burner
- Association of Official Analytical Chemistry (AOAC) 13th Edition 1980.
- Pharmacopoeia of India, British & United States.
- Hand Book of Food Analysis S. N. Mahindru.
- Analytical Biochemistry Holme Peck
- Post Graduate Chemistry Practical Part I Patel, Gadre & Turkhia.
- Agricultural Analysis. By Kanwar.
- Encyclopedia of Industrial Methods of Chemical Analysis. By F D Snell (All senus)
- Principle and practice of Analytical chemistry by F.U.Fifield and D.Keuley 3<sup>rd</sup>, Blackie and sons Ltd.
- Cosmetics by WD Poucher (Three volumes)
- Perfumery Technology (JC1) by B. Bilat and B.V.Well
- Laboratory Techniques in Food Analysis by I.M.Kolthof, D.Pearson
- Handbook of Analysis and Quality, Control for Fruits and Vegetable Products 2<sup>nd</sup> Ed by S.Ranganna
- Aids to the Analysis of Food and Drug by Nicholls
- Analysis of Food Products (Swan Publishers) by S.N.Mahendur
- Textbook of Forensic Pharmacy by B M Mithal 9th edition 1993, National Centre Kolcutta.
- Forensic Pharmacy by B.S Kuchekar, and A.M Khadatare Nirali Prakshan)

----- X ----- X -----